

d-ISOLAURELINE - A NEW ALKALOID

FROM *Liriodendron tulipifera*

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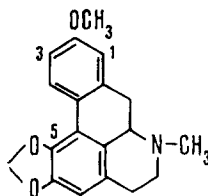
Continuing the separation of the nonphenolic fraction of the total alkaloid of the leaves of *L. tulipifera* L. (tulip tree) [1-3] on a column of silica gel, we have isolated from a benzene-methanol (98:2) eluate d-isoroemerine [2], liriodenine [3, 4], lysicamine [5], and a new alkaloid - d-isolaureline (I).

The base has the composition $C_{19}H_{19}NO_3$ (M^+ 309), hydrochloride with mp 243-245°C (decomp.), $[\alpha]_D^{24} + 35$ (c 0.25, water); R_f 0.68 in the ethyl acetate-methanol (9:1) system in a thin layer of silica gel.

The UV spectrum of (I) [λ_{max} 218, 283 nm (log ϵ +4.46, 4.26)] is characteristic for 2,5,6-trisubstituted aporphine bases [6]. The IR spectrum of the alkaloid has absorption bands at (cm^{-1}) 2850, 1255 ($-OCH_3$), 1605 (aromatic nucleus), and 1060 and 940 (OCH_2O).

The NMR spectrum of d-isolaureline shows signals (τ scale, ppm) from the protons of an $>N-CH_3$ group (7.54 ppm, 3H, singlet) and a methoxy group (6.24 ppm, 3H, singlet). Two one-proton doublets at 4.05 and 4.19 ppm ($J=1.5$ Hz) in the spectrum of the alkaloid are analogous to those observed in the NMR spectrum of dicentrine and roemerine, which are ascribed to a methylenedioxy group located in positions C_5 and C_6 of the aporphine nucleus [7].

The presence in the spectrum of (I) of two one-proton singlets at 3.58 and 3.28 ppm due to two aromatic protons in positions C_7 and C_1 and of two one-proton doublets at 2.09 and 3.20 ppm ($J=7.5$ Hz) corresponding to aromatic protons at C_4 and C_3 shows that the methoxy group is located at C_2 .



Consequently, substance (I) is d-isolaureline [8] and we are the first to have found it in a plant.

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